

ICAEEC 2014

International Conference of Asian Environmental Chemistry 2014

Programs and Abstracts



**Venue : Convention Center, Chulabhorn Research Institute
Bangkok, Thailand**

D a t e : November 24th – 26th, 2014



Japan Society for Environmental Chemistry

Outcome of ICAEC 2014

The expected outcomes are as follows;

- To declare importance to prevent pollution by micro-pollutants (declare a statement)
- To make clear environmental pollution by micro-pollutants in South East Asia, East Asia and South Pacific region
- To strengthen relationship of researchers, particularly young researchers between Thailand and Japan

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INTERACTION OF PFOS, PFOA AND 8:2 FTOH WITH HUMAN, RAT, AND MICROBIAL CYTOCHROME P450s: SIMILARITIES AND DIFFERENCES

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Keywords : PFOS, PFOA, 8:2 FTOH, mammalian and microbial cytochrome P450

Abstract : Chemicals known as perfluoroalkyl and polyfluoroalkyl substances (PFASs) do not occur naturally, but because of wide use in chemical production since the 1950s, they can be found in the environment. Currently, the two most persistent and toxic PFASs are perfluorooctanoic acid (PFOA, C₇F₁₅COOH) and perfluorooctane sulfonic acid (PFOS, C₈F₁₇SO₃H). PFOS including the form of salts and its precursor, perfluorooctane sulfonyl fluoride, are classified as persistent organic pollutant (POPs) according to Stockholm Convention on POPs. A number of studies have reported the ubiquitous distribution of PFASs in wildlife and humans [1-3] and thus it is important to determine mammalian and microbial metabolisms of PFOS and PFOA with cytochrome P450 (CYP) monooxygenases to understand distribution and impact of PFASs on human and wildlife. In the detoxification of xenobiotics, P450 enzymes play important physiological roles. For example, CYP1A1 proteins can metabolize some dioxins and PCBs by hydroxylation, but the activities of human and rat CYP1A1 proteins are very different. Recently it has been demonstrated [4] that rat CYP1A1 metabolized 3,3',4,4',5-pentachlorobiphenyl (PCB126) into 4-OH-3,3',4',5-tetrachlorobiphenyl and 4-OH-3,3',4',5,5'-pentachlorobiphenyl, but human CYP1A1 did not metabolize. Furthermore, indigenous microorganisms are a key for biotransformation and biodegradation of organic molecules such as xenobiotics, PCBs and pesticides as well as oil hydrocarbons [5-7]. Possible mechanisms of aerobic microbial biotransformation/biodegradation of fluorinated organic compounds in the polluted environment may include catalysis by fungal or bacterial CYP monooxygenases.

In order to determine behavior and metabolism of PFOS, PFOA and 8:2 FTOH, *in vitro* oxidation experiments of PFASs with rat and human CYP monooxygenases were conducted. Herbicide chlortoluron was used as positive control for CYP binding. Furthermore, indigenous microbial consortium was enriched and isolated from the environment polluted by PFASs. *In vitro* interaction of PFASs and total microbial CYP were also analyzed. These results give an insight in similarities and differences between mammalian and microbial metabolisms of PFOS, PFOA and 8:2 FTOH with CYP monooxygenases, and how ubiquitous presence of PFASs in the environment and biota affects on the metabolism.

1. Haug, L.S., et al., Environ. Int. 37, (2011) 687-693;
2. Croes K., et al., Chemosphere 89, (2012) 988-994;
3. Domingo J.L., Environ. Int. 40, (2012) 187-195;
4. Yamazaki K., et al., J. Biochem. 149, (2011) 487-494;
5. Winogradsky, S., C. R. Acad. Sci. Hebd. Seances Acad. Sci. D. 178 (1924) 1236-1239;
6. Gojgic-Cvijovic, G., et al., Biodegradation 23, (2012) 1-14;
7. Dionisi, D., ChemBioEng Rev 1, (2014) 67-82;

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Introduction

Chemicals known as perfluoroalkyl and polyfluoroalkyl substances (PFASs) do not occur naturally, but because of wide use in chemical production since the 1950s, they can be found in the environment. Currently, the two most persistent and toxic PFASs are perfluorooctanoic acid (PFOA, C₈F₁₅COOH) and perfluorooctane sulfonic acid (PFOS, C₈F₁₇SO₃H). PFOS including the form of salts and its precursor, perfluorooctane sulfonyl fluoride, are classified as persistent organic pollutants (POPs) according to Stockholm Convention on POPs. A ubiquitous distribution of PFASs in wildlife and humans is confirmed [1-3] and thus it is important to determine mammalian and microbial metabolisms of PFOS and PFOA with cytochrome P450 (CYP) monooxygenases to understand distribution and impact of PFASs on human and wildlife.

P450 enzymes play important physiological roles in the detoxification of xenobiotics. For example, CYP1A1 proteins can metabolize some dioxins and PCBs by hydroxylation, but the activities of human and rat CYP1A1 proteins are very different. Recently it has been demonstrated [4] that rat CYP1A1 metabolized 3,3',4,4',5-pentachlorobiphenyl (PCB126) into 4-OH-3,3',4',5-tetrachlorobiphenyl and 4-OH-3,3',4',5,5'-pentachlorobiphenyl, but human CYP1A1 did not metabolize. Furthermore, indigenous microorganisms are a key for biotransformation and biodegradation of organic molecules such as xenobiotics, PCBs and pesticides as well as oil hydrocarbons [5-7]. Possible mechanisms of aerobic microbial biotransformation/biodegradation of fluorinated organic compounds in the polluted environment may include catalysis by fungal or bacterial CYP monooxygenases.

Material and methods

In order to determine behavior and metabolism of PFOS, PFOA and 8:2 FTOH, *in vitro* oxidation experiments of PFASs with rat CYP monooxygenases were conducted. Yeast with transformed rat CYP1A1 gene and human CYP1A1 gene were grown in SD media (N-base 6.7 g/L; Glucose 80 g/L and Histidine 160 mg/L) and YPAD media (yeast extract 10 g/L; BactoPeptone 20 g/L; Adenin sulphate 40 mg/L and Glucose 80 g/L). Cytochrome from *Bacillus megaterium* CYP102A1 (also referred to as P450 BM-3), were used for studying interaction of PFASs with microbial cytochromes.

Hydroxylation of PFOS and PFOA with rat CYP1A1 microsomal fraction was studied: Microsomal fraction (protein concentration 40pmol), PFOS and PFOA (100 ppb) and NADPH were used. Concentration of human CYP1A1 was too low to analyze interactions.

The extracts were loaded to Solid Phase Extraction cartridge (Presep PFC-II, Wako Pure Chemical Industries) and MeOH eluates were analyzed using LC/MS/MS.

Competition for CYP binding between chlorotoluron (CT) and PFAS were studied. Herbicide chlorotoluron was used as positive control for CYP binding.

Results and discussion

The concentration of microsomal fraction based on CO difference spectroscopy of rat CYP1A1 was 0.3pmol/μL (Fig 1).

1. Products of possible hydroxylation of PFAS after incubation with CYP1A1 were studied using 6 model systems (Table 1).

Table 1. Model systems and recovery rates

	MF	NADPH	PFOS	PFOA	ppb	%
1.	+	+	+	-	80.1	80
2.	+	+	-	+	81.8	82
3.	+	-	+	-	80.6	81
4.	+	-	-	+	96.2	96
5.	-	+	+	-	82.3	82
6.	-	+	-	+	99.5	99

MF-microsomal fraction

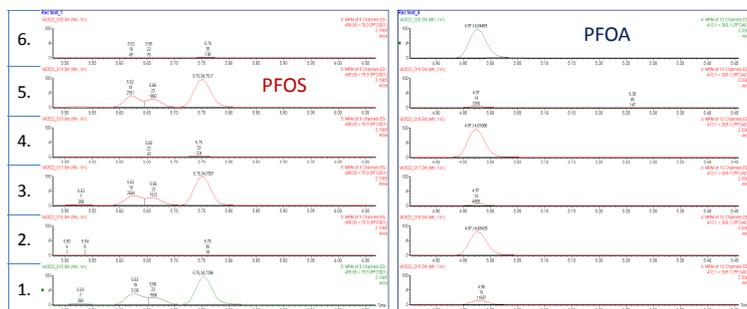
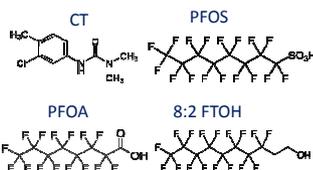


Fig 1. LC/MS chromatograms of 6 model systems extracts

However, recovery rate is suggesting that **PFOS and PFOA interact differently with rat CYP1A1 binding place**. Recovery rate of PFOS from solutions 3 and 5 indicate reduction because of abiotic factors.

High recovery rate in 4 and 6 of PFOA indicate that decrease in solution 2 is possible consequences of interaction of PFOA with MF and NADPH.

Oxidation of PFOA could lead to change of polarity and reduction of recovery rate. However, new peaks, **hydroxylation products were not detected using LC/MS/MS**.

2. Binding interaction of CT, PFOS, PFOA and 8:2 FTOH with rat CYP1A1 and microbial CYP102A1

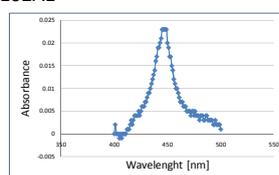


Fig 2. CO difference spectrum of rat CYP1A1

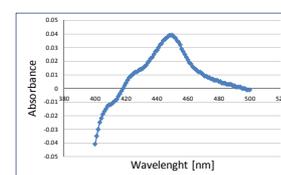


Fig 3. CO difference spectrum of microbial CYP102A1

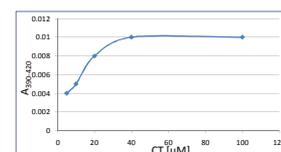
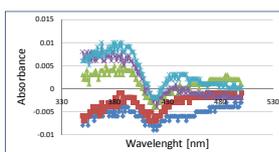


Fig 4a-b. Rat CYP1A1 spectrum of CT binding

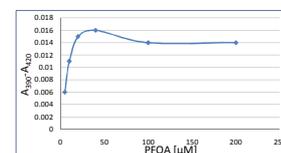
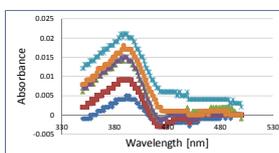


Fig 5a-b. Microbial CYP102A1 spectrum of PFOA binding

Table 2. Kd values of CT, PFOS, PFOA and 8:2 FTOH binding to rat CYP1A1 and microbial CYP102A1

Kd - rat CYP 1A1 [μM]						
CT	PFOS	PFOA	8:2 FTOH	CT & PFOS	CT & PFOA	CT & 8:2FTOH
10.2	44.2	3.94	-42.6	-1.47	0.164	-4.76
Kd - microbial CYP102A1 [μM]						
CT	PFOS	PFOA	8:2 FTOH	CT & PFOS	CT & PFOA	CT & 8:2FTOH
8.62	37.9	1.61	-0.00986	10.3	7.50	-2.32

Rat CYP1A1 Kd for **CT** is **10.2 μM**, for **PFOS** is **44.2 μM** and for **PFOA** is **3.94 μM**. Microbial CYP102A1 Kd for **CT** is **8.62 μM**, for **PFOS** is **37.9 μM** and for **PFOA** is **1.61 μM**. These results suggest that these three substrates are capable to enter binding pocket of P450 and to interact with this enzyme although CT is structurally not similar to PFOS and PFOA. Furthermore, Kd values suggest that **binding affinity increase in the following order: PFOS<CT<PFOA both in mammalian and microbial cytochroms tested**.

During incubation with 8:2 FTOH binding in the binding pocket was not determined in Rat CYP1A1 nor microbial CYP102A1. However, during incubation of Rat CYP1A1 and Microbial CYP102A1 cytochromes with CT & 8:2 FTOH it was observed that binding of CT was inhibited. In model systems with CT + PFOS and CT + PFOA it was observed that kinetics of the reaction did not followed the hyperbolic relationship between the rate of reaction and the concentration of substrate. This data suggest that there is a **competition between CT and PFOS as well as CT and PFOA for the binding pocket of P450**.

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References

- Haug, L.S., et al., Environ. Int. 37, (2011) 687-693;
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- Winogradsky, S., C. R. Acad. Sci. Hebd. Seances Acad. Sci. D. 178 (1924) 1236-1239;
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